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(54) Title: FILM COMPOSITION				
(57) Abstract				
A metallized multilayer film made from a polymer propylene polymer which can be located on a surface of	substrat the sub	e, preferably polypropylene, is modified by a maleic anhydride modified strate or incorporated within the substrate has a skin layer of polyvinyl		

alcohol which is capable of being metallized.

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FILM COMPOSITION

The invention relates to a film composition which is capable of being metallized. More specifically, the invention 5 relates to a polymeric film substrate having a polyvinyl alcohol (PVOH) skin layer which is capable of being metallized. The invention also relates to metallized films.

Oriented plastic film, specifically biaxially oriented polypropylene film is widely used for packaging products, 10 particularly foods. No single unmodified polymeric film, however, has the gas and moisture barrier characteristics needed for packaging.

The present invention is directed to a film composition capable of being metallized comprising a polymeric substrate 15 having a surface which is modified by a maleic anhydride modified propylene homopolymer or copolymer, the modified surface of the polymer substrate having a skin layer of polyvinyl alcohol.

The film composite of this invention offers excellent metal adhesion. It is resistant to flavor scalping and protects ink decals. Without primer, the film adheres well to ink and is ink printable. The film also provides excellent flavor/aroma barrier properties, lamination bond strength and optical clarity.

The multilayer film structure of this invention comprises a substrate of polyolefin, typically polypropylene. The preferred polypropylene matrix material has the following characteristics: density of 0.91 g.cc; melting point of 160°C or higher; melt flow rate of 2 to 4.

30 The polyolefin substrate is modified by a maleic anhydride modified polypropylene. The maleic anhydride modified polypropylene skin layer can be prepared by any process, for example, the process disclosed in U.S. Patent Nos. 3,433,777 and 4,198,327. A simple process for its preparation is described in U.S. 5,153,074. The maleic anhydride modified polypropylene can also be interblended into the substrate, by melt blending it with the substrate polyolefin or it can be coextruded with

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A commercially available maleic the substrate polymer. anhydride modified polypropylene or propylene copolymer has the following physical characteristics: density of 0.90 (ASTM D1505), Vicat softening point of 143°C (ASTM D1525); Shore 5 hardness of 67°C (ASTM 2240); melting point of 160°C (ASTM D2117). The maleic anhydride modified polypropylene acts as a tie layer which adheres the substrate to the PVOH. important to use the maleic anhydride modified polypropylene since without it the substrate is incompatible with the PVOH. 10 Maleic anhydride modified polypropylene is sold commercially by a variety of sources, e.g. it is sold under the tradename "ADMER" by Mitsui, "BYNEL" by DuPont, and "PLEXAR" by Quantum. Usually, if extrusion coated, the modified substrate is stretched in the machine direction 4 to 7 times and then flame 15 treated.

A layer of polyvinyl alcohol (PVOH) is applied to the surface of the modified substrate. In instances where the substrate is coextruded with the maleic anhydride modified polypropylene, it is important to apply the PVOH to the modified 20 surface, otherwise, the PVOH will have difficulty adhering to the substrate. The PVOH can be applied in a thin layer by any effective process. The layer can range in thickness from 0.01 mil to 0.20 mil. Three processes of note are 1) coextrusion of a PVOH skin and the modified substrate with the use of a 25 selector plug or multicavity die; 2) extrusion coating the PVOH layer onto the surface of the modified substrate between the machine direction orientation and the transverse direction orientation; 3) co-lamination of a thin film of PVOH onto the surface of the modified polypropylene sheet between the machine 30 direction orientation step and the transverse direction orientation step. Usually, when the PVOH layer is extrusion coated onto the substrate, the film is stretched 5 to 12 times in the transverse direction and flame treated.

For extrusion coating and co-lamination, it is preferable 35 to corona or flame treat the bonding surface of the substrate prior to application of the PVOH.

A heat sealable film can be applied to the substrate on the

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side opposite to the PVOH. This can be accomplished by coextruding a heat sealable polymer, copolymer or trimer, such as ethylene-propylene-butene, onto the substrate.

The PVOH skin is then metallized by deposition of a metal thereon. Any typical process for metallization known in the art can be employed. Typical metals contemplated are aluminum, copper, chromium, magnesium, nickel, zinc, tin, silver, gold, titanium, silicon, bismuth, or any compound containing the foregoing metals or combination thereof. An aluminum coating, which is most typically employed, can be of a thickness which yields an optical density of 1.5 to 3.5. The thickness of the aluminum coating needed to meet this optical density is, typically, from 50 to 1000 angstroms. The PVOH surface adequately adheres to metal. Although we have found that it is advantageous to surface treat, by flame or corona treatment, the PVOH prior to metallization, this is not necessary.

We discovered that extrudable grades of PVOH which can be melt processed avoid the blocking problems which are experienced when PVOH is applied through a solution coating process, as 20 described in EPA 461,772. Since PVOH is melt processed there are no blocking problems and the film can be easily wound into a roll. Also because of the extrusion process, a layer ranging from 0.01 mil to 0.20 mil can be applied, this cannot be accomplished from the solution coating processes disclosed in 25 EPA 461,772 or U.S. 5,192,620 which usually achieve a layer ranging from 0.02 to 0.06 mil. A commercial source of extrudable PVOH is Vinex resin manufactured by Air Products and Chemicals, Inc.

The advantages of using 100% PVOH in multilayer films, as disclosed herein are extensive. Since 100% PVOH is employed, the film is a better oxygen barrier than previous films made with PVOH copolymers or blends. We have also found that the surface properties provided by the PVOH are responsible for superior metal adhesion, ink printing and lamination. The surface tension of PVOH is high. Without flame or corona treatment, the surface tension is about 37 dyne/cm, as compared to the surface tension of polypropylene of 29 dyne/cm. After

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surface treatment, the surface tension of PVOH can be greater than 55 dyne/cm. Additionally, since PVOH possesses an extremely high melting and softening point, ranging from higher than 170 °C, the film exhibits no metal fracture during extrusion lamination using 10 lbs/rm LDPE at 620°F. The following Table 1 presents a comparison between the melting point and metal fracture of multi-layered polypropylene films having skin layers of EVOH, PVOH/EAA and 100% PVOH:

TABLE 1

10 POLYMER MELTING POINT, °C METAL FRACTURE
EVOH 158 little-none
PVOH/EAA* ~100 some
PVOH >170 None
* ethylene acrylic acid

15

Any grade of PVOH can be employed for the skin layer. In general partially hydrolyzed grades (of about 88% hydrolysis) can be used and even PVOH with 70% or higher hydroxyl content can be used.

The heat sealable layer employed herein and applied during coextrusion can be an ethylene propylene (EP) copolymer or an ethylene propylene butene-1 (EPB) terpolymer which, as noted above is, typically, located opposite the PVOH skin layer. The ratio of ethylene to propylene to butene-1 can be from 0 to 15% ethylene, 70 to 100% propylene and 0 to 15% butene-1; that is, 2% ethylene, 94% propylene and 4% butene-1.

A printable surface which is adhered to the metallized film by an adhesive such as low density polyethylene can be any polymer such as polyolefin homopolymer, copolymer or terpolymer, polycarbonate or polyester. The characteristic of the printable surface is one with a surface free energy of 34 dynes/cm or higher.

As demonstrated in the following Examples the PVOH in combination with the metallization and coating with an appropriate lamination film will improve oxygen barrier, moisture barrier and aroma barrier properties over other structures. While, in these examples, metallization is accomplished by conventional vacuum deposition, it can also

by conventional vapor deposition. accomplished be although polypropylene is shown as the Additionally, substrate, other polyolefin films may be used such polyesters, polybutene, olefin copolymers, polyethylene, 5 polyamides, polycarbonate and polyacrylonitrile.

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The following Examples illustrate the invention.

Example 1

This example is for comparative purposes.

A polypropylene homopolymer film was coextruded to make a 0.90 mil film, using maleic anhydride grafted polypropylene as a tie layer and an ethylene-propylene-butene terpolymer as a heat seal layer on the other side. The film was metallized on the maleic anhydride grafted polypropylene side with aluminum in a vacuum to an optical density of 2.5. The film using 10 lbs/ream LDPE hot melt. The properties of the multilayer film are set forth in Table 1.

Example 2

This example demonstrates a process for making a film in 20 accordance with the present invention which has superior air and moisture barrier properties.

coextruding polypropylene film was made by substrate with an ethylene-propylene-butene polypropylene terpolymer sealant on one side and a tie skin of maleic 25 anhydride grafted polypropylene skin on the opposite side. The film was stretched in the machine direction five times. The film was flame treated on the tie skin side. extrusion coated on the treated tie skin side and the film was stretched in the transverse direction eight times before 30 it was again flame treated on the PVOH side and wound up in a The film was vacuum metallized on the PVOH side mill roll. with aluminum to an optical density of 2.5. The aluminum coating thickness was 300 to 500 angstroms. The film was extrusion laminated with 75 gauge polypropylene film using 10 The properties of the film are set 35 lbs/ream LDPE hot melt. forth in the following Table 1.

Each film was made to a thickness of 0.90 mil. The tie

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layer (maleic anhydride grafted polypropylene) was about 0.05 mil in both examples.

Barrier properties were measured on the metallized films produced. Water vapor transmission rate (g/100 in²/24 hr) 5 was measured at 100°F, 90% RH. Oxygen transmission rate (cc/100 in²/24 hr) was measured at 73°F, 0% RH. The level of aluminum adhesion to the substrate was measured by reporting the % metal pickoff after 3 pulls on the aluminum surface with Scotch Brand 610 tape. The lamination bond (g/in) was 10 measured by laminating the metallized surface to a layer of polypropylene film of about 0.75 mil in thickness using 10 lb/ream of molten low density polyethylene at temperature of 325 °C in a laminating machine. obtained was then pulled apart to measure the lamination bond 15 strength in a conventional pulling machine. lamination bond strength testing machines are the Alfred Suter Tester and the Sintech Tensile Tester. Properties obtained are as follows:

are

	ion <u>in</u>		
	Lamination Bond, g/in	80 171	
	Metal <u>Pickoff,</u> %	8 0 IS	
	$\overline{\text{WVTR}^2}$	10.0 0.15 0.024 0.06	
Table 2	OTR ¹	10.0	
	Treatment Level, dyne/cm	40 > 56	oxygen transmission rate water vapor transmission rate
	Metallizable <u>Skin</u>	PP (Control) PVOH (6 ga)	oxygen transmis water vapor tra
	EX.	2 1	$10 \frac{1}{2}$

film the As shown in Table 1 the oxygen barrier and moisture barrier properties of excellent and the metal-to-film adhesion is appropriate for most purposes.

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CLAIMS

- A film composition which comprises a polymeric substrate having at least one surface which is modified by a
 maleic anhydride modified propylene homopolymer or copolymer, the modified surface of the polymer substrate having a skin layer, capable of being metallized, consisting essentially of polyvinyl alcohol.
- 2. A film composition according to claim 1 in which the maleic anhydride modified propylene homopolymer or copolymer is a discrete layer.
- 3. A film composition according to claim 1 or 2 in 15 which the substrate comprises an oriented homopolymer or copolymer of propylene.
 - 4. A film composition according to any preceding claim in which one side of the film has a heat seal layer thereon.

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- 5. A film composition according to any preceding claim in which the polyvinyl alcohol skin layer is essentially free from crosslinking agent.
- 25 6. A film composition according to any preceding claim in which the polyvinyl alcohol skin layer comprises extrudable polyvinyl alcohol.

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- 7. A film composition of claim 11 in which the polyvinyl alcohol skin layer is a partially hydrolyzed grade.
- 8. A metallized film composition according to any 5 preceding claim in which the skin layer consisting of essentially polyvinyl alcohol has a metal layer directly thereon.
- 9. A metallized film composition according to claim 8
 10 in which there is a heat sealable layer on the substrate opposite the skin layer.

INTERNATIONAL SEARCH REPORT

International application No. PCT/US95/06619

A. CLASSIFICATION OF SUBJECT MATTER IPC(6) :B32B 27/08,≠5/04, 27/00, 1/08; CO8F 255/02 US CL :428/515, 518, 520, 522, 457, 461, 463, 910, 35.9, 36.6, 36.7 According to International Patent Classification (IPC) or to both national classification and IPC				
B. FIELDS SEARCHED				
Minimum documentation searched (classification system follower	d by classification symbols)			
U.S. : 428/515, 518, 457, 461, 35.9, 36.6, 36.7				
Documentation searched other than minimum documentation to the	e extent that such documents are included in the fields searched			
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Electronic data base consulted during the international search (n	ame of data base and, where practicable, search terms used)			
C. DOCUMENTS CONSIDERED TO BE RELEVANT				
Category* Citation of document, with indication, where a	ppropriate, of the relevant passages Relevant to claim No.			
X US, A, 5,153,074 (MIGLIORINI) 0 column 1, line 49 to column 3, lin				
A US, A, 4,181,689 (NAGATOSHI see entire document.	ET AL) 01 January 1980, 1-3			
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Further documents are listed in the continuation of Box	C. See patent family annex.			
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INTERNATIONAL SEARCH REPORT

International application No. PCT/US95/06619

Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)				
his international report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:				
Claims Nos.: - because they relate to subject matter not required to be searched by this Authority, namely:				
Claims Nos.: because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:	1			
Claims Nos.: 4-9 because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).				
lox II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)				
This International Searching Authority found multiple inventions in this international application, as follows:				
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As all required additional search fees were timely paid by the applicant, this international search report covers all search claims.	hable			
As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite pay of any additional fee.	ment			
As only some of the required additional search fees were timely paid by the applicant, this international search report conly those claims for which fees were paid, specifically claims Nos.:	overs			
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No required additional except for over timely said by the applicant. Consequently, this intermetical except was				
No required additional search fees were timely paid by the applicant. Consequently, this international search represented to the invention first mentioned in the claims; it is covered by claims Nos.:	ole is			
Remark on Protest				
No protest accompanied the payment of additional search fees.	1			

Form PCT/ISA/210 (continuation of first sheet(1))(July 1992)★